Regioselective Synthesis of Enones via a Titanium-Promoted Coupling of Unsymmetrical Alkynes with Weinreb Amides

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S Supporting Information

[AB](#page-6-0)STRACT: [A modular tit](#page-6-0)anium-promoted coupling of unsymmetrical internal alkynes with Weinreb amides is described. The coupling reaction takes place at room temperature and affords Etrisubstituted enones in moderate to good yields with high levels of regioselectivity. The system shows moderate chemoselectivity.

α,β-Unsaturated ketones (enones) are important building blocks that have found widespread use in the synthesis of natural products¹ and biologically active small molecules.² The utility of enones is in the diverse array of structures that can be accessed from t[hi](#page-6-0)s motif. Thus, numerous methods hav[e](#page-6-0) been developed for the preparation of enones, $3 \text{ such as dehydrostatic}$ aldol reactions,⁴ oxidations,⁵ olefination reactions with α ketophosphonates, 6 palladium dehydro[ge](#page-6-0)nation, 7 transitionmetal-catalyzed [is](#page-6-0)omerizatio[n](#page-6-0) of propargylic alcohols,⁸ hydroacylation of alk[yn](#page-6-0)es,⁹ and acylation of or[ga](#page-6-0)nometallic reag[en](#page-6-0)ts.¹⁰ While all of these methods complement each other, there is still a n[ee](#page-6-0)d for a method that can be used to prepare [tri](#page-6-0)- and tetrasubstituted enones in a modular manner from readily available and easily modifiable substrates.

As part of our endeavors to develop methodologies for applications in natural product mimic libraries, we speculated that a titanium-promoted coupling¹¹ of alkynes and acyl electrophiles could afford enones in a modular manner with high selectivity. Stemming from [t](#page-6-0)he seminal work of Kulinkovich on the cyclopropanation of esters, 12 Sato demonstrated that diisopropoxytitanacyclopropenes undergo intramolecular nucleophilic acyl substitution with ca[rb](#page-6-0)onates (Scheme 1a).¹³ Six expanded this work to include the formation of α , β -unsaturated carboxylic acids and esters via an intermole[cul](#page-6-0)ar titanium-mediated reductive coupling of alkynes with carbon dioxide¹⁴ and carbonates (Scheme 1b).¹⁵ Complementarily, zirconocene-mediated couplings of alkynes with chloroformates¹⁶ and [ca](#page-6-0)rbon dioxide¹⁷ have also be[en](#page-6-0) reported. Surprisingly, the use of group IV metallacycles to form enones from [a](#page-6-0)lkynes has not b[een](#page-6-0) reported. For applications in library preparation, a vast pool of readily available acyl electrophiles are needed to facilitate diversification of the library. On the basis of this requirement, we chose to investigate Weinreb amides, which are bench-stable reactive acyl electrophiles that can easily and efficiently be prepared from ubiquitous carboxylic acids.¹⁸ Weinreb amides have been utilized in enone synthesis with vinyllithium¹⁹ and Grignard²⁰ reagents but not with titana- or [zir](#page-6-0)cona-cyclopropenes.

Evaluation of the titanacycle literature i[nd](#page-6-0)icated that t[he](#page-6-0) optimal starting point would be generation of the titanacyclopropene via Sato's method, which involves reduction of

a) Sato's intramolecular nucleophilic acyl substitution

b) Six's alkyne-carbonate coupling

c) This work

 $Ti(OiPr)₄$ with *i-PrMgCl* followed by ligand exchange with an alkyne.²¹ A limitation of this method for applications in small-molecule library synthesis is that diisopropoxytitanacyclopropen[es](#page-6-0) are thermally unstable and typically cannot be warmed above −30 °C. It was speculated that the rate of reaction of the titanacyclopropene with a Weinreb amide would be faster than the decomposition pathways and side reactions at room temperature and, in addition, that the coupling would generate a stabilized intermediate that would be thermally stable through coordination of the methoxy group to the titanium.

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While the ultimate goal was the development of a simple mix-and-stir procedure that would operate at room temperature, the reaction was first investigated under cryogenic reaction conditions. A room-temperature procedure would require the titanacyclopropene to be generated in the presence of the Weinreb amide, and it was not clear whether the Grignard reagent would preferentially react with $Ti(OiPr)₄$ over the Weinreb amide. Thus, for the initial optimization reactions, the titanacyclopropene of 1-phenyl-1-propyne (10) was generated under prototypical cryogenic conditions (−78 to −40 °C for 3 h), after which the Weinreb amide (11) was added at $-78~^\circ\mathrm{C}$ and the reaction mixture was warmed to room temperature. Under these conditions the desired enone was formed (Table 1, entry 1), establishing that a titanacyclopro-

Table 1. Optimization of the Titanium-Promoted Coupling Conditions^a

	Ph-		.Me Ph OMe	1.5 equiv Ti(OiPr) ₄ 3.0 equiv reducing agent	
		Me $\ddot{}$		Solvent	Ph R_1 R_{2}
	10		11		12a, R ₁ = Ph & R ₂ = Me 12b, R_1 = Me & R_2 = Ph
	entry	solvent	equiv of 11	reducing agent	% yield ^b $(12a/12b)^c$
	1	Et ₂ O	1	i-PrMgCl	32 ($>99/1$)
	$\overline{2}$	Et ₂ O	2	i-PrMgCl	59(97/3)
	3	Et ₂ O	5	i-PrMgCl	52(97/3)
	$\overline{4}$	THF	$\overline{2}$	i-PrMgCl	34
	5	MTBE	2	i-PrMgCl	Ω
	6	toluene	2	i-PrMgCl	$\mathbf{0}$
	7	Et ₂ O	2	EtMgBr	Ω
	8	Et ₂ O	1	c -C ₅ H ₉ MgCl	20
	\mathfrak{g}^d	THF	2	n-BuLi	37(14/86)
	10^{ef}	THF	2	n-BuLi	Ω
	11 ^g	Et ₂ O	2	i-PrMgCl	58 (97/3)
	12^h	Et ₂ O	2	i-PrMgCl	67(97/3)

^aConditions: alkyne 10 (1 mmol), $Ti(OiPr)₄$ (1.5 mmol), reducing agent (3 mmol), and Et₂O (10 mL) were mixed at -78 °C; the mixture was warmed to -40 °C for 3 h and then cooled to -78 °C; Weinreb amide 11 was added; the mixture was warmed to room temperature. ^bIsolated yields after flash chromatography. ^cRatios of regioisomers were determined by GC−MS analysis of the crude reaction mixtures. d_n -BuLi was added at -78 °C, the mixture was warmed to room temperature, and then Weinreb amide 11 was added. n-BuLi was added last at room temperature. ^f1-Phenylpentanone was isolated in 84% yield. g _i-PrMgCl was added last at -78 °C, and then the mixture was warmed to room temperature. h_i -PrMgCl was added the mixture was warmed to room temperature. last at room temperature

pene can undergo coupling with a Weinreb amide with high regioselectivity, favoring enone 12a. Increasing the amount of Weinreb amide 11 to 2 equiv increased the yield of the enone (entry 2), but further increases had no effect (entry 3). 22 Attempts to further increase the yield through adjustment of the solvent (entries 4–6) established Et₂O as the opti[mal](#page-6-0) choice. The choice of reducing agent was critical, as a negative effect was seen with EtMgBr (entry 7) and cyclopentylmagnesium chloride (entry 8). It has been demonstrated that the use of n -BuLi produces thermally stable titanacycles.²³ Surprisingly, under this reaction manifold the use of n -BuLi to generate the titanacyclopropene afforded the enone in mod[era](#page-6-0)te yield but with a complete reversal in regioselectivity, favoring enone 12b (entry 9). Simplification of the procedure by generating the titanacyclopropene in the presence of the Weinreb amide was

explored next. It was found that addition of n -BuLi to the reaction mixture last did not result in titanacycle formation; instead, the n-BuLi selectively reacted with only the Weinreb amide to afford 1-phenylpentanone (entry 10). Conversely, addition of i-PrMgCl to the reaction mixture last at −78 °C followed by warming to room temperature (entry 11) afforded the desired enone in the same yield and regioselectivity as the stepwise process (entry 2). The yield was increased further to 67% by running this sequential addition of reagents at room temperature (entry 12).

Having established the optimal conditions to be the operationally simple procedure of combining all of the reagents and then adding the i-PrMgCl last at room temperature followed by stirring for 4 h, substrate screening was initiated to determine the scope of this coupling reaction (Table 2). Initial efforts focused on varying the Weinreb amide while keeping the alkyne static, using the unsymmetrical alkyne 10. In [al](#page-2-0)l of the cases examined (entries 1−22), the regioselectivity of the coupling reaction was greater than 97/3. Formation of the carbon−carbon bond favored the alkyne carbon that contained the sterically smaller substituent, regardless of the Weinreb amide employed. Aliphatic Weinreb amides (entries 12−16) were higher-yielding than aromatic Weinreb amides (entries 1− 11), except for the sterically large N-methoxy-N-methylpivalamide (entry 17), which afforded the enone in a moderate yield comparable to those for aromatic counterparts. The system tolerated aromatic and aliphatic halogens (entries 2−6 and 13), although it is noteworthy that an aromatic bromide did not undergo magnesium halogen exchange with the Grignard reagent.²⁴ The system tolerated *meta*- and *para*-substituted aromatic Weinreb amides, whereas ortho-substituted²⁵ aromatics [\(I,](#page-6-0) OMe, Me) would not react with the in situ-formed titanacycle (entries 7 and 10), with the exceptio[n](#page-6-0) of a naphthalene (entry 11). Methoxy groups inhibited the coupling (entry 8), presumably because of coordination with the titanium complex. To establish that coordination was the issue and not deactivation of the Weinreb amide carbonyl by electron donation, a Weinreb amide with a TBS-protected phenol was prepared and reacted without incident (entry 9). Additional functional groups on the aromatic Weinreb amide that were found to be incompatible with the system were $-NO₂$, −CN, −OAc, and −C(O)CH₃. Furan (entry 18) and thiophene (entry 19) heteroaromatics were tolerated, but a pyridine (entry 20) inhibited coupling.²⁶ Conjugated Weinreb amides underwent the coupling to produce the corresponding 1,4-dien-3-ones in modest yields (ent[rie](#page-6-0)s 21 and 22). These products have the potential to undergo a Nazarov cyclization,² but under these reaction conditions less than 3% yield of the Nazarov product was observed.

A nearly quantitative yield was obtained with 4-octyne (entry 23), whereas sterically more congested diphenylacetylene (entry 24) afforded the enone in moderate yield. As already demonstrated with 1-phenyl-1-propyne, an alkyne with a clear steric difference between the groups attached to the alkyne undergoes the coupling with high regioselectivity. To test the limits of the regioselectivity, unsymmetrical alkynes of varying steric bulk were examined. The coupling reaction afforded a near 1/1 mixture of regioisomers when the steric differentiating group was distal to the alkyne (entry 25) or when the two groups were similarly sized (entry 27). A single isomer was obtained with 3-benzyloxy-1-propynylbenzene (entry 26), albeit in low yield, presumably as a result of the formation of a dimeric titanacycle that inhibited the reaction with the

1.5 equiv Ti(O/Pr)₄

 \overline{O}

Table 2. Substrate Screening of the Titanium-Mediated Coupling of Alkynes and Weinreb Amides^{a,b} Ω

^aConditions: alkyne (1 mmol), Weinreb amide (2 mmol), Ti(OiPr)₄ (1.5 mmol), Et₂O (10 mL), *i*-PrMgCl (3 mmol, added last), room temperature, 4 h. ^b Unless otherwise stated, the ratio of regioisomers was ≥97/3 as determined by GC−MS analysis of the crude reaction mixture; the major isomer is shown. Collated yield by flash chromatography. ^d 41% of the alkyne was recovered. ^e 52% of the alkyne was recovered.

Weinreb amide. High regioselectivity was seen with silylprotected terminal alkynes (entries 29−31). Sterically congested phenyl(tert-butyl)acetylene participated in the coupling, affording the enone in moderate yield but with high selectivity (entry 32). Increasing the sterics further with tert-butyl- (trimethylsilyl)acetylene completely inhibited the reaction.²⁸

Our earlier findings demonstrated that when n-BuLi was used as the reducing agent (Table 1, entry 9), the enone was for[me](#page-6-0)d favoring the opposite regioisomer. On the basis of this result, we became interested in the [p](#page-1-0)ossibility of selectively accessing either regioisomer simply by changing the reducing agent. Unfortunately, when n -BuLi was employed in the coupling reaction the yields were considerably lower, and with a number of the substrates screened there was no reaction. Control reactions were run to help determine which element was the cause of the reversal in regioselectivity. First, we explored the possibility of a solvent effect. Since the solution of i-PrMgCl used was in ether and the n-BuLi was in hexanes, we ran the coupling reaction with the Grignard reagent in the presence of added hexanes. This resulted only in a lower yield of enone with no erosion in the selectivity, consistent with the solvent screening data, where the reaction did not take place in toluene (Table 1, entry 6). Next we examined the role of the metal cation. To examine whether the lithium cation interacted with the tita[niu](#page-1-0)m complex and thereby affected the regiocontrol, 10 equiv of LiCl was added to the reaction with i-PrMgCl and was found to have no effect. In theory, reduction of the $Ti(OiPr)_4$ with the reducing agent produces 2 equiv of a metal alkoxide, which could affect the aggregate structure of the titanacycle and/or coordinate to the titanium to form an ate complex. Thus, we ran the coupling reaction using *i*-PrMgCl with 2 equiv of lithium tert-butoxide added, which lowered the yield of the enone from 92% to 66% and slightly eroded the regiocontrol to 96/4. The root cause of the regioselectivity difference with n -BuLi and i-PrMgCl remains unclear.

Titanacycles are carbanion reagents, so we examined subsequent reactions of the stabilized titanacycle intermediate. As shown in Scheme 2, the in situ-formed titanacycle 14 could

Scheme 2. Subsequent Reactions of Stabilized Intermediate Titanacycle 14

be quenched with deuterium oxide to afford the deuterated enone in 82% yield with greater than 95% deuterium incorporation, and titanium halogen exchange of 14 with iodine afforded the conjugated vinyl iodide in moderate yield. Attempts to transmetalate the chelated and stabilized titanium− carbon bond in titanacycle 14 with copper(I) salts were successful only with CuOtBu, albeit in low yield, but this could be used to couple with allyl bromide to form the skipped dienone 17. The reaction of titanacycle 14 with benzaldehyde did not occur because of the decreased reactivity of the stabilized titanacycle, but precomplexation of the aldehyde with BF_3 ·OEt₂ facilitated the addition and subsequent cyclization to yield tetrasubstituted furan 18 in 66% yield.²⁹ A more detailed examination of this approach to furan and heterocycle synthesis is underway and will be reported in due co[urs](#page-6-0)e.

In summary, this report has described a titanium-mediated coupling of internal alkynes with Weinreb amides to yield Etrisubstituted enones in moderate to good yields. The regioselectivity of the reaction is due to the steric difference between the groups attached to the alkyne, with levels as high >99/1 being obtained. Additionally, this is the first demonstration that organotitanium reagents react with Weinreb amides, thereby expanding the arsenal of nucleophiles that can react with this important acyl electrophile.

EXPERIMENTAL SECTION

Methods. All of the reactions were carried out in oven-dried or flame-dried glassware under an atmosphere of argon with magnetic stirring. Reactions were monitored either by thin-layer chromatography with 0.25 mm precoated silica gel plates or by gas chromatography. Visualization of all TLCs was performed with UV light and/or staining with phosphomolybdic acid, KMnO₄, or Seebach's stain. Purifications were performed by flash chromatography with silica gel (60 Å, 230−400 mesh) packed in glass columns and elution with hexanes/EtOAc, unless otherwise noted.

Materials. Diethyl ether, dichloromethane, chloroform, and tetrahydrofuran were purified and dried using a solvent purification system that contained activated alumina. 1,2-Dichloroethane and pyridine were freshly distilled from calcium hydride under argon. All of the Weinreb amides were prepared following literature procedures³⁰ from purchased carboxylic acids.

Instrumentation. ¹H NMR and ¹³C NMR spectra were obtain[ed](#page-7-0) on a 400 MHz NMR spectrometer (400 MHz for ¹H and 101 MHz for 13 C), and chemical shifts (δ , ppm) are reported relative to residual chloroform solvent peaks (δ = 7.26 ppm for ¹H and δ = 77.0 ppm for chloroform solvent peaks (δ = 7.26 ppm for ¹H and δ = 77.0 ppm for ¹²H NMR are reported as follows: chemical shift

(multiplicity, coupling constant(s) in Hz, integration). Multiplicities are indicated as follows: $s =$ singlet, $d =$ doublet, $t =$ triplet, $q =$ quartet, $quin = quintet, sex = sextet, sept = septet, m = multiplet or unresolved.$ The reported melting points are uncorrected. Low-resolution mass spectra were obtained by GC−MS, and high-resolution mass spectrometry (HRMS) was performed using an Orbitrap operated in FT mode to provide a nominal resolution of 100 000.

General Procedure. A round-bottom flask was sealed with a septum, and the system was placed under an atmosphere of argon by performing a vacuum-purge cycle three times and then attaching a balloon of argon. The round-bottom flask was charged with the alkyne (1 mmol), Weinreb amide (2 mmol), dry diethyl ether (10 mL), and titanium isopropoxide (1.5 mmol, 0.44 mL). To this stirring mixture was injected a solution of isopropylmagnesium chloride (2 M in ether, 3 mmol, 1.5 mL) dropwise over 5 min. The reaction mixture was stirred at room temperature for 4 h, after which the system was opened to the air and the mixture was quenched with 1 mL of water. The mixture was dried over magnesium sulfate, filtered, and concentrated. The crude material was subjected to flash chromatography, eluting with hexanes/EtOAc (98/2), unless otherwise noted. Note 1: any solid reagents were added to the flask prior to the vacuum-purge cycle. Note 2: a small aliquot of the quenched reaction mixture was used to determine the regioisomer ratio by GC−MS.

(E)-2-Methyl-1,3-diphenylprop-2-en-1-one (Table 2, entry 1): subjection of 1-phenyl-1-propyne (1 mmol, 0.125 mL) and Nmethoxy-N-methylbenzamide (2.0 mmol, 0.304 mL) to the general procedure afforded 0.149 g (67% yield) of the enone a[s](#page-2-0) a yellow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.81–7.75 (m, 3H), 7.59−7.52 (m, 1H), 7.51−7.32 (m, 7H), 7.21 (q, J = 1.2 Hz, 1H), 2.30 (d, J = 1.2 Hz, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 199.3, 142.1, 138.4, 136.7, 135.6, 131.5, 129.6, 129.4, 128.5, 128.4, 128.1, 14.3. The physical and spectral data were consistent with those reported in the literature. $3¹$

(E)-2-Methyl-3-phenyl-1-(4-fluorophenyl)prop-2-en-1-one (Table 2, entry 2): subjection o[f 1](#page-7-0)-phenyl-1-propyne (1.0 mmol, 0.125 mL) and 4-fluoro-N-methoxy-N-methylbenzamide (2.0 mmol, 0.311 mL) to the general procedure afforded 0.145 g (60% yield) of the enone as [a](#page-2-0) yellowish liquid after flash chromatography. ^IH NMR (400 MHz, CDCl₃) δ 7.80 (dd, J = 8.6, 5.5 Hz, 2H), 7.42 (m, 5H), 7.14 (m, 3H), 2.27 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 197.9, 164.9, 163.6, 141.7, 136.6, 135.5, 134.4, 132.0, 131.9, 129.6, 128.6, 128.4, 115.4, 115.3, 14.5; IR (neat) 3055, 2923, 1644, 1594, 1258, 1225, 1154, 1010, 691, 637 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₃FO 241.1024, found 241.1020.

(E)-1-(4-Chlorophenyl)-2-methyl-3-phenylprop-2-en-1-one (Table 2, entry 3): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and 4-chloro-N-methoxy-N-methylbenzamide (2.0 mmol, 0.326 mL) to the general procedure afforded 0.145 g (57% yield) of the enone as [a](#page-2-0) yellow oil after flash chromatography. 1H NMR (400 MHz, CDCl₃) δ 7.70 (m, 2H), 7.42 (m, 7H), 7.14 (m, 1H), 2.26 (s, 3H); 13C NMR (101 MHz, CDCl₃) δ 198.0, 142.2, 137.9, 136.7, 136.5, 135.4, 130.8, 129.6, 128.7, 128.4, 128.4, 14.4. The physical and spectral data were consistent with those reported in the literature.³²

(E)-1-(3-Bromophenyl)-2-methyl-3-phenylprop-2-en-1-one (Table 2, entry 4): subjection of 1-phenyl-1-propyne [\(1.](#page-7-0)0 mmol, 0.125 mL) and 3-bromo-N-methoxy-N-methylbenzamide (2.0 mmol, 0.334 mL) to the general procedure afforded 0.171 g (57% yield) of the enone as [a](#page-2-0) brownish liquid after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.87 (m, 1H), 7.66 (d, J = 8.4 Hz, 2H), 7.45–7.30 (m, 6H), 7.18 (s, 1H), 2.27 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 197.6, 142.9, 140.4, 136.4, 135.3, 134.4, 132.1, 129.7, 128.8, 128.4, 128.2, 127.8, 122.4, 14.2; IR (neat) 3058, 2923, 1645, 1562, 1248, 1018, 728, 695 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₃BrO 301.0223, found 301.0221.

(E)-2-Methyl-3-phenyl-1-(4-(trifluoromethyl)phenyl)prop-2-en-1 one (Table 2, entry 5): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and 4-trifluoromethyl-N-methoxy-N-methylbenzamide (2.0 mmol, 0.369 mL) to the general procedure afforded 0.159 g (55% yi[e](#page-2-0)ld) of the enone as a yellow liquid after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.82 (d, J = 8.0 Hz, 2H), 7.73 (d, J = 8.4 Hz, 2H), 7.42 (m, 5H), 7.17 (s, 1H), 2.28 (s, 3H); 13C NMR (101 MHz, CDCl₃) δ 198.1, 143.7, 141.9, 136.5, 135.3, 133.1, 132.7, 129.8, 129.5, 128.9, 128.5, 125.2, 125.2, 125.2, 125.1, 14.0; IR (neat) 3055, 2962, 1649, 1616, 1321, 1107, 1064, 1022, 693 cm[−]¹ ; HRMS (ESI) m/ z [M + H]⁺ calcd for C₁₇H₁₃F₃O 291.0992, found 291.0992.

(E)-2-Methyl-3-phenyl-1-(3-trifluoromethyl)phenyl)prop-2-en-1 one (Table 2, entry 6): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and 3-trifluoromethyl-N-methoxy-N-methylbenzamide (2.0 mmol, 0.369 mL) to the general procedure afforded 0.161 g (56% yi[e](#page-2-0)ld) of the enone as a yellow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 8.01 (s, 1H), 7.92 (d, J = 7.6 Hz, 1H), 7.80 (d, J = 8.4 Hz, 1H), 7.60 (t, J = 7.6 Hz, 1H), 7.43 (m, 5H), 7.17 (s, 1H), 2.30 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 197.8, 143.2, 139.2, 136.5, 135.3, 132.5, 130.9, 130.6, 129.7, 128.9, 128.8, 128.5, 128.07, 128.03, 128.00, 127.9, 126.17, 126.13, 126.09, 126.05, 14.2; IR (neat) 2926, 1647, 1611, 1575, 1331, 1244, 1093, 1071, 695 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₃F₃O 291.0992, found 291.0994.

(E)-1-(4-((tert-Butyldimethylsilyl)oxy)phenyl)-2-methyl-3-phenylprop-2-en-1-one (Table 2, entry 9): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and TBS-protected 4-hydroxy-N-methoxy-Nmethylbenzamide (2.0 mmol, 0.529 mL) to the general procedure afforded 0.189 g (53% y[iel](#page-2-0)d) of the enone as a yellowish oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.77 (d, J = 8.8 Hz, 2H), 7.42 (m, 5H), 7.14 (s, 1H), 6.91 (d, J = 8.8 Hz, 2H), 2.27 (s, 3H), 1.02 (s, 9H), 0.27 (s, 6H); ¹³C NMR (101 MHz, CDCl₃) δ 198.2, 159.3, 140.0, 136.8, 135.8, 131.8, 131.2, 129.5, 128.2, 119.5, 29.8, 25.5, 18.1, 14.8, −4.4; IR (neat) 2955, 2928, 2857, 1643, 1595, 1505, 1253, 906, 837, 805, 736, 691 cm[−]¹ ; HRMS (ESI) m/z [M + $[H]^+$ calcd for $C_{22}H_{28}O_2Si$ 353.1932, found 353.1936.

(E)-2-Methyl-1-(naphthalen-1-yl)-3-phenylprop-2-en-1-one (Table 2, entry 11): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and N-methoxy-N-methyl-1-naphthamide (2.0 mmol, 0.430 g) to the general procedure afforded 0.055 g (20% yield) of the enone as a [ye](#page-2-0)llow oil after flash chromatography. 1 H NMR (400 MHz, CDCl3) δ 8.01 (d, $J = 8.0$ Hz, 1H), 7.98 (d, $J = 8.0$ Hz, 1H), 7.93 m, 1H), 7.53 (m, 4H), 7.40−7.29 (m, 5H), 7.22 (s, 1H), 2.37 (s, 3H); 13C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 200.8, 144.9, 138.5, 137.4, 135.6, 133.6, 131.0, 130.2, 129.8, 128.8, 128.4, 128.3, 126.9, 126.4, 126.3, 125.6, 124.4, 13.4; IR (neat) 3054, 2922, 1643, 1574, 1245, 1197, 777, 692 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₁₆O 273.1274, found 273.1270.

(E)-2-Methyl-1,5-diphenylpent-1-en-3-one (Table 2, entry 12): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and Nmethoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.217 g (87% yield) of [th](#page-2-0)e enone as a yellow oil after flash chromatography. ${}^1\mathrm{H}$ NMR (400 MHz, CDCl3) δ 7.50 (s, 1H), 7.44–7.19 (m, 10H), 3.15 (t, J = 8.0 Hz, 2H), 3.02 (t, J = 8.0 Hz, 2H), 2.08 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 201.3, 141.5, 138.7, 137.2, 135.8, 129.7, 128.47, 128.42, 128.40, 126.1, 39.6, 30.8, 13.2. The physical and spectral data were consistent with those reported in the literature.³³

(E)-5-Chloro-2-methyl-1-phenylpent-1-en-3-one (Table 2, entry 13): subjection of 1-phe[nyl](#page-7-0)-1-propyne (1.0 mmol, 0.125 mL) and 3 chloro-N-methoxy-N-methylpropanamide (2.0 mmol, 0.266 mL) to the general procedure afforded 0.153 g (74% yield) of the e[no](#page-2-0)ne as a yellow oil after flash chromatography $(r.r. = 97/3)$. ¹H NMR (400 MHz, CDCl₃) δ 7.51 (s, 1H), 7.43 (m, 5H), 3.87 (t, J = 6.4 Hz, 2H), 3.31 (t, J = 6.9 Hz, 2H), 2.08 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 198.4, 139.4, 136.9, 135.4, 129.6, 128.7, 128.4, 40.2, 39.2, 12.9; IR (neat) 2962, 2919, 1661, 1575, 1195, 1065, 727, 694, 657 cm⁻¹; HRMS (ESI) m/z : [M + H]⁺ calcd for C₁₂H₁₃ClO 209.0728, found 209.0729.

(E)-4-Ethyl-2-methyl-1-phenyloct-1-en-3-one (Table 2, entry 14): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and 2-ethyl-N-methoxy-N-methylhexanamide (2.0 mmol, 0.410 mL) to the general procedure afforded 0.196 g (80% yield) of the enone as a [y](#page-2-0)ellow liquid after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.51 (s, 1H), 7.43 (m, 4H), 7.33 (m, 1H), 3.26 (quin, $J = 6.8$ Hz, 1H), 2.08 (s, 3H), 1.70 (m, 2H), 1.58−1.41 (m, 2H), 1.35−1.18 (m, 4H), 0.88 (m,

6H); ¹³C NMR (101 MHz, CDCl₃) δ 206.9, 138.5, 137.9, 136.1, 129.6, 128.3, 128.3, 46.5, 32.4, 29.8, 26.0, 22.9, 13.9, 13.4, 12.0; IR (neat) 2958, 2928, 2858, 1658, 1047, 694 cm[−]¹ ; HRMS (ESI) m/z [M $+ H$]⁺ calcd for C₁₇H₂₄O 245.1900, found 245.1900.

(E)-1-Cyclohexyl-2-methyl-3-phenylprop-2-en-1-one (Table 2, entry 15): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and N-methoxy-N-methyl-1-cyclohexanamide (2.0 mmol, 0.335 mL) to the general procedure afforded 0.217 g (95% yield) of the enone [as](#page-2-0) a yellow liquid after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.49 (s, 1H), 7.40 (m, 5H), 3.16 (t, J = 11.7 Hz, 1H), 2.04 (s, 3H), 1.82 (d, J = 10.8 Hz, 4H), 1.53–1.19 (m, 6H); ¹³C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 205.8, 137.6, 136.4, 136.0, 129.5, 128.2, 128.2, 44.5, 29.8, 25.83, 25.80, 13.4; IR (neat) 2927, 2852, 1658, 1448, 1590, 1005, 753, 696 cm⁻¹; HRMS (ESI) *m*/z [M + H]⁺ calcd for C₁₆H₂₀O 229.1587, found 229.1587.

(E)-2,5,5-Trimethyl-1-phenylhex-1-en-3-one (Table 2, entry 16): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and Nmethoxy-N,3,3-trimethylbutanamide (2.0 mmol, 0.340 mL) to the general procedure afforded 0.198 g (92% yield) of t[he](#page-2-0) enone as a yellow liquid after flash chromatography. ¹ H NMR (400 MHz, CDCl₃) δ 7.47 (s, 1H), 7.37 (m, 5H), 2.71 (s, 2H), 2.04 (s, 3H), 1.06 (s, 9H); 13C NMR (101 MHz, CDCl3) δ 202.6, 139.0, 138.6, 136.1, 129.57, 128.4, 128.3, 49.2, 31.4, 30.1, 13.2; IR (neat) 2953, 2866, 1655, 1362, 764, 698 cm⁻¹; HRMS (ESI) calcd for C₁₅H₂₀O [M + H]⁺ 217.1587, found 217.1585.

(E)-2,4,4-Trimethyl-1-phenylpent-1-en-3-one (Table 2, entry 17): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and Nmethoxy-N-methylpivalamide (2.0 mmol, 0.310 mL) to the general procedure afforded 0.106 g (52% yield) of the enone [as](#page-2-0) a yellowish liquid after flash chromatography. ${}^{1}\textrm{H}$ NMR (400 MHz, CDCl₃) δ 7.37 (m, 5H), 6.93 (s, 1H), 2.09 (s, 3H), 1.36 (s, 9H); 13C NMR (101 MHz, CDCl₃) δ 211.9, 137.6, 135.9, 131.8, 129.1, 128.2, 127.5, 44.1, 27.9, 16.0; IR (neat) 2966, 2869, 1683, 1659, 1477, 1047, 765, 694 cm⁻¹; HRMS (ESI) m/z $[M + H]^+$ calcd for C₁₄H₁₈O 203.1431, found 203.1429.

(E)-1-(Furan-2-yl)-2-methyl-3-phenylprop-2-en-1-one (Table 2, entry 18): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and N-methoxy-N-methylfuran-2-carboxamide (2.0 mmol, 0.268 mL) to the general procedure afforded 0.111 g (53% yield) of the enone [as](#page-2-0) a yellow oil after flash chromatography. 1H NMR (400 MHz, CDCl₃) δ 7.66 (s, 1H), 7.50 (s, 1H), 7.46–7.30 (m, 5H), 7.17 (d, J = 3.2 Hz, 1H), 6.55 (d, $J = 3.6$ Hz, 1H), 2.23 (s, 3H); ¹³C NMR (101 MHz, CDCl3) δ 185.1, 151.8, 146.7, 139.3, 136.3, 135.7, 129.6, 128.4, 128.3, 119.5, 111.7, 14.5; IR (neat) 2916, 2848, 1630, 1575, 1558, 1463, 1389, 1273, 1024, 1011, 889, 761, 706, 692 cm⁻¹; HRMS (ESI) m/z $[M + H]^{+}$ calcd for $C_{14}H_{12}O_2$ 213.0911, found 213.0910.

(E)-2-Methyl-3-phenyl-1-(thiophen-2-yl)prop-2-en-1-one (Table 2, entry 19): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and N-methoxy-N-methylthiophene-2-carboxamide (2.0 mmol, 0.281 mL) to the general procedure afforded 0.117 g (52% yield) of t[he](#page-2-0) enone as a yellow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.69 (d, J = 3.6 Hz, 1H), 7.67 (d, J = 4.8 Hz, 1H), 7.48–7.32 (m, 6H), 7.14 (t, J = 4.4 Hz, 1H), 2.26 (s, 3H); ¹³C NMR (101 MHz, CDCl3) δ 190.5, 143.4, 139.0, 136.9, 135.6, 133.5, 133.4, 129.5, 128.4, 128.3, 127.6, 14.8; IR (neat) 2917, 1618, 1512, 1411, 1263, 1002, 847, 723, 693 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₄H₁₂OS 229.0682, found 229.0679.

 (E,E) -2-Methyl-1-phenylhexa-1,4-dien-3-one (Table 2, entry 21): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and (E)-Nmethoxy-N-methylbut-2-enamide (2.0 mmol, 0.264 mL) to the general procedure afforded 0.056 mg (31% yield) of the enone [as](#page-2-0) a yellowish liquid after flash chromatography. ¹H NMR (400 MHz, CDCl3₃) δ 7.46 (m, 1H), 7.44−7.30 (m, 5H), 6.95 (dq, J = 15.2, 6.8 Hz, 1H), 6.79 $(dq, J = 15.2, 1.49 \text{ Hz}, 1H), 2.11 (s, 3H), 1.95 (d, J = 6.8 \text{ Hz}, 3H);$ ¹³C NMR (101 MHz, CDCl₃) δ 192.8, 143.0, 138.4, 137.9, 135.9, 129.6, 128.3, 128.2, 126.9, 18.3, 13.5; IR (neat) 3024, 2913, 1659, 1612, 1575, 1491, 1441, 1287, 1206, 1063, 964, 915, 753, 694 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₃H₁₄O 187.1118, found 187.1116.

(E,E)-2-Methyl-1,5-diphenylpenta-1,4-dien-3-one (Table 2, entry 22): subjection of 1-phenyl-1-propyne (1.0 mmol, 0.125 mL) and N- methoxy-N-methylcinnamamide (2.0 mmol, 0.382 g) to the general procedure afforded 0.066 g (27% yield) of the enone as a yellow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.72 (d₃ J $= 15.6$ Hz, 1H), 7.62 (m, 3H), 7.50–7.33 (m, 9H), 2.20 (s, 3H); ¹³C NMR (101 MHz, CDCl₃) δ 192.6, 143.4, 138.6, 138.5, 135.9, 135.1, 130.1, 129.7, 128.8, 128.5, 128.4, 128.2, 121.9, 13.8; IR (neat) 3025, 2920, 1650, 1593, 1494, 1448, 1328, 1200, 1061, 762, 698 cm⁻ ; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₆O 249.1274, found 249.1272.

(E)-1-Phenyl-4-propyloct-4-en-3-one (Table 2, entry 23): subjection of 4-octyne (1.0 mmol, 0.147 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.232 g (95% yield) of the enone as a [ye](#page-2-0)llow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.28 (m, 2H), 7.21 $(m, 3H)$, 6.57 $(t, J = 7.3$ Hz, 1H), 2.97 $(m, 4H)$, 2.27 $(m, 2H)$, 2.21 $(q,$ $J = 7.4$ Hz, 2H), 1.47 (sex, $J = 7.2$ Hz, 2H), 1.33 (sex, $J = 7.2$ Hz, 2H), 0.95 (t, $J = 7.2$ Hz, 3H), 0.89 (t, $J = 7.2$ Hz, 3H); ¹³C NMR (101) MHz, CDCl₃) δ 200.7, 142.8, 141.8, 141.6, 128.39, 128.38, 125.9, 39.2, 30.9, 30.8, 27.7, 22.5, 22.2, 14.2, 13.9. The physical and spectral data were consistent with those reported in the literature.³⁴

(E)-1,2,5-Triphenylpent-1-en-3-one (Table 2, entry 24): subjection of diphenylacetylene (1.0 mmol, 0.178 g) and N-met[hox](#page-7-0)y-N-methyl-3 phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.149 g (48% yield) of the enone a[s](#page-2-0) a colorless solid after flash chromatography. $\dot{Mp} = 88 \text{ °C}$; ¹H NMR (400 MHz, CDCl₃) δ 7.69 (s, 1H), 7.44 (m, 3H), 7.36−7.13 (m, 10H), 7.06 (m, 2H), 3.00 $(t, J = 7.2 \text{ Hz}, 2H), 2.93 \text{ (t, } J = 6.8 \text{ Hz}, 2H);$ ¹³C NMR (101 MHz, CDCl3) δ 200.4, 141.3, 140.4, 138.2, 136.8, 134.6, 130.8, 129.5, 129.1, 129.0, 128.39, 128.38, 128.2, 127.9, 125.9, 41.8, 30.4; IR (neat) 3027, 2922, 1676, 1568, 1353, 1281, 1190, 738, 696 cm^{−1}; HRMS (ESI) *m/z* $[M + H]^{+}$ calcd for $C_{23}H_{20}O$ 313.1587, found 313.1584.

(E)-7-((tert-Butyldimethylsilyl)oxy)-4-ethyl-1-phenylhept-4-en-3 one (major isomer) and (E) -4- $(2-((\text{tert-butyldimethylsilyl)oxy})$ ethyl)-1-phenylhept-4-en-3-one (minor isomer) (Table 2, entry 25): subjection of tert-butyl(hex-3-yn-1-yloxy)dimethylsilane (1.0 mmol, 0.252 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded [0](#page-2-0).259 g (75% yield) of an inseparable mixture of these regioisomeric enones (r.r. = 53/47) as a colorless liquid after flash chromatography. Major isomer: ¹ ¹H NMR (400 MHz, CDCl₃) δ 7.21 (m, 2H), 7.13 (m, 3H), 6.59 (t, J $= 7.2$ Hz, 1H, measured 0.43H), 3.51 (t, $J = 6.8$ Hz, 2H, measured 0.89H), 2.89 (m, 4H), 2.49 (t, $J = 6.8$ Hz, 2H, measured 0.87H), 2.27 (m, 2H), 0.99 (t, J = 7.6 Hz, 3H, measured 1.37H), 0.82 (s, 9H, measured 3.64H), −0.03 (s, 6H, measured 2.40H); 13C NMR (101 MHz, CDCl₃) δ 200.4, 146.4, 141.54, 137.6, 128.38, 128.33, 125.94, 62.2, 39.13, 32.3, 30.6, 29.4, 25.9, 18.3, 13.4, −5.4. Minor isomer: ¹H NMR (400 MHz, CDCl₃) δ 7.21 (m, 2H), 7.13 (m, 3H), 6.54 (t, J = 7.2 Hz, 1H, measured 0.49H), 3.65 (t, J = 6.4 Hz, 2H, measured 1.04H), 2.89 (m, 4H), 2.40 (q, J = 6.8 Hz, 2H, measured 1.09H), 2.24 (m, 2H), 0.88 (t, J = 7.6 Hz, 3H, measured 1.96H), 0.83 (s, 9H, measured 5.04H), 0.00 (s, 6H, measured 3.29H); 13C NMR (101 MHz, CDCl₃) δ 200.3, 144.4, 141.49, 138.7, 128.35, 128.32, 125.91, 61.8, 39.10, 30.7, 25.8, 22.4, 19.0, 18.2, 13.8, −5.4. IR (neat) 2955, 2928, 2856, 1668, 1496, 1471, 1251, 1094, 833, 774, 697 cm^{−1}; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₃₄O₂Si 347.2401, found 347.2391.

(E)-2-((Benzyloxy)methyl)-1,5-diphenylpent-1-en-3-one (Table 2, entry 26): subjection of (3-(benzyloxy)prop-1-yn-1-yl)benzene (1.0 mmol, 0.206 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.045 g (1[3%](#page-2-0) yield) of the enone as a yellow oil after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.69 (s, 1H), 7.48 (m, 2H), 7.39-7.19 $(m, 13H)$, 4.59 (s, 2H), 4.35 (s, 2H), 3.12 (t, J = 8.0 Hz, 2H), 3.01 (t, J $= 8.0$ Hz, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 200.2, 143.4, 141.4, 137.9, 136.8, 134.7, 129.8, 129.4, 128.5, 128.48, 128.43, 128.3, 128.2, 127.7, 126.1, 73.1, 63.6, 40.0, 30.4; IR (neat) 3026, 2924, 1668, 1494, 1452, 1069, 1027, 733 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{25}H_{24}O_2$ 357.1850, found 357.1837.

(E)-2-(Cyclohex-1-en-1-yl)-1,5-diphenylpent-1-en-3-one (isomer 1) and (E)-1-(cyclohex-1-en-1-yl)-2,5-diphenylpent-1-en-3-one (isomer 2) (Table 2, entry 27): subjection of (cyclohex-1-en-1-ylethynyl)- benzene (1.0 mmol, 0.189 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.191 g (60% yield) of an inseparable mixture of these regioisomeric enones (r.r. = 50/50) as a yellow oil after flash chromatography. Isomer 1: ¹ H NMR (400 MHz, CDCl3) δ 7.62 (m, 2H), 7.42−7.15 (m, 9H), 5.66 (m, 1H), 3.05 (m, 4H), 2.19 (m, 2H), 1.75 (m, 4H), 1.51 (m, 2H); ¹³C NMR (101 MHz, CDCl₃) δ 201.10, 142.40, 141.48, 137.6, 136.3, 135.7, 135.2, 130.2, 128.6, 128.3, 128.2, 127.8, 125.9, 40.6, 30.6, 28.3, 25.4, 22.3, 21.7. Isomer 2: ¹H NMR (400 MHz, CDCl3) δ 7.42−7.15 (m, 11H), 6.22 (m, 1H), 2.92 (m, 2H), 2.84 (m, 2H), 2.11 (m, 4H), 1.65 (q, J = 7.6 Hz, 2H), 1.41 (m, 2H); ¹³C NMR $(101 \text{ MHz}, \text{CDCl}_3)$ δ 200.6, 142.8, 141.5, 140.8, 136.6, 134.1, 130.1, 129.0, 128.5, 128.4, 128.3, 127.4, 125.8, 41.6, 30.5, 27.2, 26.8, 22.5, 21.5. IR 3025, 2929, 1668, 1571, 1494, 1446, 1179, 1071, 908, 729 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₂₄O 317.1900, found 317.1890.

(E)-4-(Cyclohex-1-en-1-ylmethylene)-1-phenyldecan-3-one (major isomer) and (E)-4-(cyclohex-1-en-1-yl)-1-phenylundec-4-en-3-one (minor isomer) (Table 2, entry 28): Subjection of 1-(oct-1-yn-1 yl)cyclohex-1-ene (1.0 mmol, 0.219 mL) and N-methoxy-N-methyl-3 phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.208 g (64% [yi](#page-2-0)eld) of an inseparable mixture of these regioisomeric enones (r.r. = $77/23$) as a yellowish liquid after flash chromatography. Major isomer: ^{1}H NMR (400 MHz, CDCl₃) δ 7.27− 7.05 (m, 5H), 6.75 (s, 1H, measured 0.58H), 5.85 (m, 1H, measured 0.56H), 2.89 (m, 4H), 2.35 (t, $J = 7.2$ Hz, 2H, measured 1.74H), 2.10 $(m, 4H, measured 3.69H), 1.57 (m, 4H), 1.21 (m, 8H), 0.81 (t, J = 6.8$ Hz, 3H). Minor isomer: ¹H NMR (400 MHz, CDCl₃) δ 7.27−7.05 (m, 5H), 6.57 (t, J = 7.6 Hz, 1H, measured 0.19H), 6.53 (s, 1H, measured 0.22H), 2.83 (m, 2H), 2.69 (m, 2H, measured 0.41H), 2.35 (m, 2H, measured 1.74H), 1.86 (m, 4H, measured 0.75H), 1.57 (m, 4H), 1.21 (m, 8H), 0.81 (t, $I = 6.8$ Hz, 3H). Combined isomers: ¹³C NMR (101 MHz, CDCl₃) δ 201.5, 141.9, 141.6, 139.3, 135.0, 133.6, 128.3, 127.1, 125.9, 39.5, 31.5, 30.9, 30.2, 29.5, 28.2, 26.6, 26.1, 22.6, 22.5, 21.7, 14.0. IR (neat) 3026, 2924, 2855, 1663, 1495, 1452, 1105, 921, 747, 697 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₃₂O 325.2526, found 325.2517.

(E)-1,7-Diphenyl-4-((trimethylsilyl)methylene)heptan-3-one (Table 2, entry 29): subjection of trimethyl(5-phenylpent-1-yn-1 yl)silane (1.0 mmol, 0.240 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.180 g [\(](#page-2-0)51% yield) of the enone as a yellowish liquid after flash chromatography (r.r. = $91/9$; only a single isomer was isolated). ¹H NMR (400 MHz, CDCl₃) δ 7.16 (m, 4H), 7.07 (m, 6H), 6.42 (s, 1H), 2.88 (t, $J = 8.0$ Hz, 2H), 2.80 (t, $J = 7.6$ Hz, 2H), 2.52 (t, $J = 7.6$ Hz, 2H), 2.28 (t, $J = 8.0$ Hz, 2H), 1.52 (quin, $J = 7.6$ Hz, 2H), 0.00 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 201.3, 156.1, 142.0, 141.4, 139.9, 128.4, 128.39, 128.37, 128.2, 126.0, 125.7, 39.4, 36.3, 31.9, 31.0, 30.6, −0.4; IR (neat) 3025, 2951, 1672, 1602, 1495, 1452, 1248, 836, 745, 696 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₃H₃₀OSi 351.2139, found 351.2137.

(E)-2-(Cyclohex-1-en-1-yl)-5-phenyl-1-(trimethylsilyl)pent-1-en-3 one (Table 2, entry 30): subjection of (cyclohex-1-en-1-ylethynyl) trimethylsilane (1.0 mmol, 0.207 mL) and N-methoxy-N-methyl-3 phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.1[43](#page-2-0) g (46% yield) of the enone as a yellow oil after flash chromatography that was an inseparable mixture of regioisomers $(r.r. =$ 86/14). ¹H NMR (400 MHz, CDCl₃) δ 7.32 (m, 2H), 7.24 (m, 3H), 6.57 (bs, 1H, measured 0.02H), 6.52 (bs, 1H, measured 0.75H), 5.62 (m, 1H, measured 0.18H), 5.53 (tt, $J = 3.6$, 1.7 Hz, 1H, measured 0.77H), 2.97 (m, 4H), 2.15 (m, 2H), 2.03 (m, 2H), 1.69 (m, 4H), 0.12 (s, 9H, measured 7.69H), −0.04 (s, 9H, measured 1.29H); 13C NMR (101 MHz, CDCl3) δ 201.4, 158.8, 141.5, 139.5, 137.4, 128.4, 127.4, 126.0, 40.4, 30.5, 29.0, 25.1, 22.4, 21.7, −0.2; IR (neat) 3027, 2928, 1674, 1496, 1245, 836, 748, 697 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for $C_{20}H_{28}OSi$ 313.1983, found 313.1978.

(E)-2,5-Diphenyl-1-(trimethylsilyl)pent-1-en-3-one (Table 2, entry 31): subjection of 1-phenyl-2-trimethylsilylacetylene (1.0 mmol, 0.197 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.149 g (53% [yie](#page-2-0)ld) of the enone as a yellow liquid after flash chromatography that was an inseparable mixture of regioisomers (r.r. = $86/14$). ¹H NMR (400 MHz, CDCl3) δ 7.50−7.15 (m, 10H), 7.11 (m, 1H, measured 0.92H), 3.04 (m, 4H, measured 3.18H), 3.00−2.75 (m, 4H, measured 0.89H), 0.00 (bs, 9H, measured 6.96H), −0.37 (bs, 9H, measured 1.37H); ¹³C NMR (101 MHz, CDCl₃) δ 200.1, 155.4, 142.3, 141.3, 138.7, 129.3, 128.4, 128.4, 127.9, 127.7, 125.9, 41.1, 30.4, −0.7; IR (neat) 3027, 2952, 1679, 1578, 1247, 1099, 856, 834, 747, 697 cm[−]¹ ; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₂₄OSi 309.1670, found 309.1667.

(E)-6,6-Dimethyl-1,4-diphenylhept-4-en-3-one (Table 2, entry 32): subjection of (3,3-dimethylbut-1-yn-1-yl)benzene (1.0 mmol, 0.180 mL) and N-methoxy-N-methyl-3-phenylpropanamide (2.0 mmol, 0.365 mL) to the general procedure afforded 0.104 g ([35](#page-2-0)% yield) of the enone as a colorless liquid after flash chromatography. ¹H NMR (400 MHz, CDCl₃) δ 7.35 (m, 3H), 7.30–7.18 (m, 4H), 7.13 (m, 3H), 6.89 (s, 1H), 2.89 (t, $J = 8.0$ Hz, 2H), 2.75 (t, $J = 8.0$ Hz, 2H), 0.94 (s, 9H); ¹³C NMR (101 MHz, CDCl₃) δ 200.8, 151.1, 141.4, 139.5, 137.0, 130.0, 128.3, 128.2, 127.8, 127.3, 125.8, 44.8, 41.5, 34.1, 30.4; IR (neat) 3026, 2959, 1688, 1593, 1495, 1475, 1359, 1215, 1111, 1072, 1030, 747 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₂₄O 293.1900, found 293.1890.

■ ASSOCIATED CONTENT

6 Supporting Information

 1 H and 13 C NMR spectra for all of the products. This material is available free of charge via the Internet at http://pubs.acs.org.

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Notes

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■ REFERENCES

(1) Ueda, A.; Yamamoto, A.; Kato, D.; Kishi, Y. J. Am. Chem. Soc. 2014, 136, 5171−5176.

(2) Ding, C.; Zhang, Y.; Chen, H.; Yang, Z.; Wild, C.; Ye, N.; Ester, C. D.; Xiong, A.; White, M. A.; Shen, Q.; Zhou, J. J. Med. Chem. 2013, 56, 8814−8825.

(3) For reviews of enone synthesis, see: (a) Marsden, S. P. Sci. Synth. 2005, 26, 1045−1121. (b) Buckle, D. R.; Pinto, I. L. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, U.K., 1991; Vol. 7, pp 119−149. (c) Larock, R. C. Comprehensive Organic Transformations, 2nd ed.; John Wiley & Sons: New York, 1999; pp 251−256.

(4) Heathcock, C. H. In Comprehensive Organic Synthesis; Trost, B. M., Fleming, I., Eds.; Pergamon: Oxford, U.K., 1991; Vol. 2, pp 133− 179.

(5) (a) Ren, K.; Hu, B.; Zhao, M.; Tu, Y.; Xie, X.; Zhang, Z. J. Org. Chem. 2014, 79, 2170−2177. (b) Nakano, T.; Ishii, Y.; Ogawa, M. J. Org. Chem. 1987, 52, 4855−4859. (c) Liu, J.; Ma, S. Org. Lett. 2013, 15, 5150−5153. For an example of allylic oxidation, see: (d) Catino, A. J.; Forslund, R. E.; Doyle, M. P. J. Am. Chem. Soc. 2004, 126, 13622−13623. For examples of oxidative dehydrogenation, see: (e) Nicolaou, K. C.; Zhong, Y. L.; Baran, P. S. J. Am. Chem. Soc. 2000, 122, 7596−7597.

(6) Corey, E. J.; Weinshenker, N. M.; Schaaf, T. K.; Huber, W. J. Am. Chem. Soc. 1969, 91, 5675−5677.

(7) Muzart, J. Eur. J. Org. Chem. 2010, 3779−3790.

(8) For an example with Au, see: (a) Pennell, M. N.; Unthank, M. G.; Turner, P.; Sheppard, T. D. J. Org. Chem. 2011, 76, 1479−1482. For an example with Ag, see: (b) Sugawara, Y.; Yamada, W.; Yoshida, S.; Ikeno, T.; Yamada, T. J. Am. Chem. Soc. 2007, 129, 12902−12903. For an example with Ru, see: (c) Cadierno, V.; Crochet, P.; Garcia-Garrido, S. E.; Gimeno, J. Dalton Trans. 2010, 39, 4015−4031. For examples with Cu, see: (d) Collins, B. S. L.; Suero, M. G.; Gaunt, M. J. Angew. Chem., Int. Ed. 2013, 52, 5799−5802. (e) Xiong, Y.-P.; Wu, M.- Y.; Zhang, X.-Y.; Ma, C.-L.; Huang, L.; Zhao, L.-J.; Tan, B.; Liu, X.-Y. Org. Lett. 2014, 16, 1000−1003.

(9) (a) For a review of transition-metal-catalyzed hydroacylation, see: Willis, M. C. Chem. Rev. 2010, 110, 725−748. (b) Biju, A. T.; Wurz, N. E.; Glorius, F. J. Am. Chem. Soc. 2010, 132, 5970−5971. (c) Parsons, S. R.; Hooper, J. F.; Willis, M. C. Org. Lett. 2011, 13, 998−1000.

(10) For an example with organolithium reagents, see: (a) Kong, K.; Moussa, Z.; Lee, C.; Romo, D. J. Am. Chem. Soc. 2011, 133, 19844− 19856. For an example with organomagnesium reagents, see: (b) Fu, R.; Ye, J.-L.; Dai, X.-J.; Ruan, Y.-P.; Huang, P.-Q. J. Org. Chem. 2010, 75, 4230−4243. For an example with organosilicon reagents, see: (c) Fleming, I.; Pearce, A. J. Chem. Soc., Perkin Trans. 1 1980, 2485− 2489. For an example with organostannanes, see: (d) Echavarren, A. M.; Stille, J. K. J. Am. Chem. Soc. 1988, 110, 1557−1565. For an example with organoboranes, see: (e) Ishiyama, N.; Miyaura, N.; Suzuki, A. Bull. Chem. Soc. Jpn. 1991, 64, 1999−2001.

(11) (a) Wolan, A.; Six, Y. Tetrahedron 2010, 66, 15−61. (b) Wolan, A.; Six, Y. Tetrahedron 2010, 66, 3097−3133.

(12) Kulinkovich, O. G.; de Meijere, A. Chem. Rev. 2000, 100, 2789− 2834.

(13) (a) Kasatkin, A.; Okamoto, S.; Sato, F. Tetrahedron Lett. 1995, 36, 6075−6078. (b) Okamoto, S.; Kasatkin, A.; Zubaidha, P. K.; Sato, F. J. Am. Chem. Soc. 1996, 118, 2208−2216.

(14) Six, Y. Eur. J. Org. Chem. 2003, 1157−1171.

(15) Wolan, A.; Cadoret, F.; Six, Y. Tetrahedron 2009, 65, 7429− 7439.

(16) Takahashi, T.; Xi, C.; Ura, Y.; Nakajima, K. J. Am. Chem. Soc. 2000, 122, 3228−3229.

(17) Yamashita, K.; Chatani, N. Synlett 2005, 919−922.

(18) (a) Nahm, S.; Weinreb, S. M. Tetrahedron Lett. 1981, 22, 3815− 3818. For a review of Weinreb amides, see: (b) Balasubramaniam, S.; Aidhen, I. S. Synthesis 2008, 3707−3738.

(19) Bach, J.; Galobardes, M.; Garcia, J.; Romea, P.; Tey, C.; Urpi, F.; Vilarrasa, J. Tetrahedron Lett. 1998, 39, 6765−6768.

(20) Yu, L.-F.; Hu, H.-N.; Nan, F.-J. J. Org. Chem. 2011, 76, 1448− 1451.

(21) (a) Harada, K.; Urabe, H.; Sato, F. Tetrahedron Lett. 1995, 36, 3203−3206. For a review of titanium alkyne complexes, see: (b) Sato, F.; Urabe, H.; Okamoto, S. Chem. Rev. 2000, 100, 2835−2886.

(22) It was determined later that for aliphatic Weinreb amides as little as 1.2 equiv could be used to obtain the optimal yield.

(23) (a) Rassadin, V.; Six, Y. Tetrahedron 2014, 70, 787−794. (b) Obora, Y.; Moriya, H.; Tokunaga, M.; Tsuji, Y. Chem. Commun. 2003, 2820−2821. (c) Eisch, J. J.; Gitua, J. N. Organometallics 2003, 22, 24−26.

(24) Knochel, P.; Dohle, W.; Gommermann, N.; Kneisel, F. F.; Kopp, F.; Korn, T.; Sapauntzis, I.; Vu, V. A. Angew. Chem., Int. Ed. 2003, 42, 4302−4320.

(25) For a similar trend in the reaction of aromatic Weinreb amides, see: Rudzinski, D. M.; Kelly, C. B.; Leadbeater, N. E. Chem. Commun. 2012, 48, 9610−9612.

(26) For an example of a pyridine ligand on a titanium complex, see: Noor, A.; Kempe, R. Eur. J. Inorg. Chem. 2008, 2377−2381.

(27) Spencer, W. T., III; Vaidya, T.; Frontier, A. J. Eur. J. Org. Chem. 2013, 3621−3633.

(28) The alkyne was recovered in nearly quantative yield, and titanacycle formation did not occur.

(29) (a) Suzuki, D.; Nobe, Y.; Watai, Y.; Tanaka, R.; Takayama, Y.; Sato, F.; Urabe, H. J. Am. Chem. Soc. 2005, 127, 7474−7479. (b) Teng,

The Journal of Organic Chemistry Note

X.; Wada, T.; Okamoto, S.; Sato, F. Tetrahedron Lett. 2001, 42, 5501− 5503.

- (30) (a) Rudzinski, D. M.; Kelly, C. B.; Leadbeater, N. E. Chem. Commun. 2012, 48, 9610−9612. (b) Shintani, R.; Kimura, T.; Hayashi, T. Chem. Commun. 2005, 3213−3214.
- (31) Niu, T.; Zhang, W.; Huang, D.; Xu, C.; Wang, H.; Hu, Y. Org. Lett. 2009, 11, 4474−4477.
- (32) Wei, Y.; Tang, J.; Cong, X.; Zeng, X. Green Chem. 2013, 15, 3165−3169.
- (33) Fujihara, T.; Tatsumi, K.; Terao, J.; Tsuji, Y. Org. Lett. 2013, 15, 2286−2289.
- (34) Ooi, I.; Sakurai, T.; Takaya, J.; Iwasawa, N. Chem.-Eur. J. 2012, 18, 14618−14621.